



ONE-POT CATALYTIC VALORIZATION OF BIOMASS TO ETHYLENE GLYCOL OVER GLUCOSE-DERIVED CARBON-BASED CATALYSTS

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Catalytic conversion of biomass is highly attractive for producing high added-value products, being ethylene glycol (EG) one of the most highlighted ^[1]. However, its production usually requires expensive metals (e.g., Ru) and carbon supports (e.g., carbon nanotubes (CNT), activated carbon). This work aimed to develop less expensive carbon-supported metal catalysts for the direct conversion of biomass to EG.

Glucose-based carbon materials were prepared by HTC. The materials were then carbonized (CG) or physically activated (AG_x) (Table 1). Ni-W catalysts were prepared by incipient wetness impregnation of the supports, and the one-pot conversion of cellulose/wastes to EG was performed in a reactor at 205 °C and 50 bar of H₂. The prepared Ni-W catalysts were in general highly efficient, with 100 % cellulose conversion (*X*) (Table 1). Ni-W/AG₁₀₀₀ was the most efficient: EG yield (Y_{EG}) up to 60 %. These results surpassed previous works using Ru-W supported on CNT ^[2] or glucose-based materials ^[3], indicating that both CNT and Ru can be successfully replaced by low-cost alternatives. The best catalyst is also being evaluated for the conversion of wastes (e.g., paper, food waste), which so far resulted in EG yields up to 50 %. Thus, these materials are herein presented as low-cost and sustainable catalysts.

| | Support | Gas type and flow rate | T (°C) | <i>t</i> (h) | X (%) | Yeg (%) |
|--|---------|--|--------|--------------|-------|---------|
| | CG | N ₂ , 50 cm ³ min ⁻¹ | 700 | 2 | 100 | 51.6 |
| | AG600 | CO ₂ , 80 cm ³ g ⁻¹ min ⁻¹ | 700 | 2 | 100 | 41.8 |
| | AG1000 | CO ₂ , 80 cm ³ g ⁻¹ min ⁻¹ | 900 | 2 | 100 | 59.5 |
| | AG2200 | CO ₂ , 80 cm ³ g ⁻¹ min ⁻¹ | 900 | 6 | 100 | 56.3 |

Table 1. Experimental conditions of the materials and catalytic results after 5 h.

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References:

[1] Baniamerian, H.; Hoj, M.; Beier, M.J.; Jensen, A.D. Appl. Catal. B: Environ., 2023, 330, 122650.

[2] Ribeiro, L.S.; Órfão, J.J.M.; Pereira, M.F.R. Bioresour. Technology, 2018, 263, 402.

[3] Ribeiro, L.S.; Rey-Raap, N.; Figueiredo, J.L.; Órfão, J.J.M.; Pereira, M.F.R. Cellulose, 2019, 26, 7337.